

UNIVERSITI TUN HUSSEIN ONN MALAYSIA

FINAL EXAMINATION SEMESTER II SESSION 2013/2014

COURSE NAME

: CHEMICAL PROCESS AND

SUSTAINABILITY

COURSE CODE

: BNQ 20603

PROGRAMME

: 2 BNN

TEST DATE

: JUNE 2014

DURATION

: 3 HOURS

INSTRUCTION

: ANSWER FOUR (4)

QUESTIONS ONLY

THIS QUESTION PAPER CONSISTS OF FOUR (4) PAGES

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- Q1 Catalytic reagents are known superior to stoichiometric reagents.
 - (a) Explain the **THREE** (3) general characteristics of catalyst.

(9 marks)

(b) Catalyst makes the chemical process more efficient in terms of energy consumption. Sketch a diagram to assist your answer.

(8 marks)

- (c) State TWO (2) types of biocatalyst and for each biocatalyst list:
 - (i) **ONE** (1) advantage
 - (ii) TWO (2) disadvantages

(8 marks)

- Q2 Chemical industries contribute to disposal of industrial such as DDT and pollute the environment. The chemicals do not readily degrade and persist in environment (air, water, soil).
 - (a) Analyze and discuss which of the Green Chemistry Principles apply to the above statement and give further definition of the principle.

(10 marks)

(b) One of the method of onsite waste treatment is biotreatment plants. Explain the purpose of biotreatment plant and the concept of biodegradation.

(8 marks)

(c) Chemicals waste which are highly toxic, need appropriate control and maintain. Explain the Ames method for measuring the potential harmful effects of chemicals.

(7 marks)

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Q3	Yo	ou are a technologist working in the development of a new industr	ial process of	
	Polylactic acid (PLA) plastic. The lead chemist in the project mentions that solvent that			
		Il be used for the process is acetone.		
	(a)	Give FOUR (4) caharacteristics of acetone.		
			(8 marks)	
	(b)	Explain FOUR (4) characteristic of suitable alternative solvent to replacetone?	ace the	
			(8 marks)	
	(c)	Plan the projects to be more environmentally friendly by: (i) Suggesting the alternatives solvents that will replace acetone.		
		(ii) Predict TWO (2) advantages of the alternatives solvents.	(9 marks)	
Q4	Lactic acid, ethanol and acetic acid are bio-based chemicals produced from renewable resources. Choose only two of these bio-chemicals and answer the following question:			
	(a)	Analyze and discuss which of the Green Chemistry Principles apply to statement.	the above	
			(10 marks)	
	(b)	State ONE (1) type of bioprocess methods to produce each of the chose chemicals.	en bio-	
			(5 marks)	
	(c)	Outline TWO (2) examples of:		
	, ,	(i) Derivatives produced from the process	(5 marks)	
		(ii) Applications of the bio-based chemicals	(5 marks)	

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Q5	Benzyl alcohol (10.81g, 0.10 mol, FW108.1) is reacted with p toluenesulfong (21.9 g, 0.115 mol, FW 190.65) in solution [toluene (500 g) and triethylamine induce the reaction to give the sulfonate ester (FW 262.29) isolated in 90% mol, 23.6 g). Calculate for the above reaction:	e (15 g)] to
	(a) The atom economy (AE),	
		(5 marks)
	(b) The reaction mass efficiency (RME),	
		(5 marks)
	(c) The mass intensity (MI),	
		(5 marks)
	(d) The mass productivity (MP).	
		(5 marks)
	(e) The <i>E</i> -factor.	
		(5 marks)

- END OF QUESTION -